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**INVESTIGATION of Si and SiO₂ ETCH MECHANISMS
USING an INTEGRATED SURFACE KINETICS MODEL**

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AGENDA

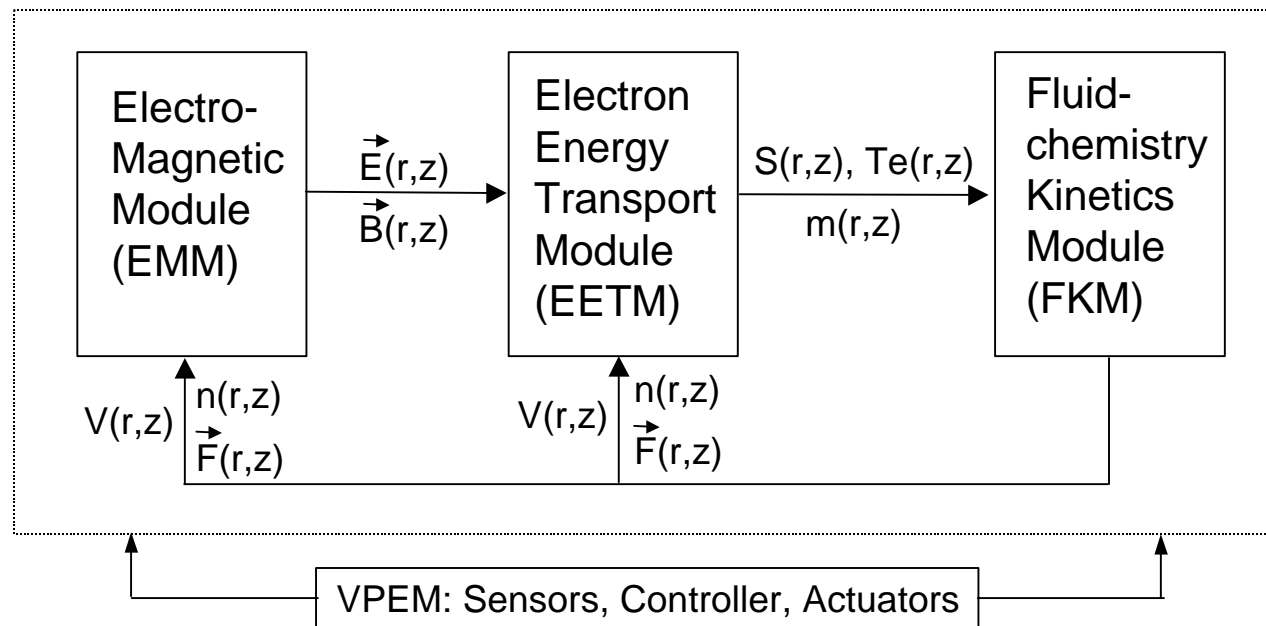
- Introduction
- Description of the Model
 - Hybrid Plasma Equipment Model (HPEM)
 - Surface Kinetics Module (SKM)
- C₂F₆ etching of Si
 - Reactor wall temperature
- C₂F₆ etching of SiO₂
 - Substrate bias
- Summary

INTRODUCTION

- Fluorocarbon (C_mF_n) plasmas are widely used for Si/SiO₂ etching due to their favorable kinetic properties and high selectivity.
- CF_x radicals passivate the wafer surface and so influence etching behavior.
- The passivation thickness depends on wafer materials, plasma-wall interactions, and processing parameters.
- Our goal:
 - Develop a surface reaction model
 - Couple the surface model with the bulk plasma simulator - HPEM
 - Quantitatively investigate the surface reaction mechanisms
 - Optimize the processes

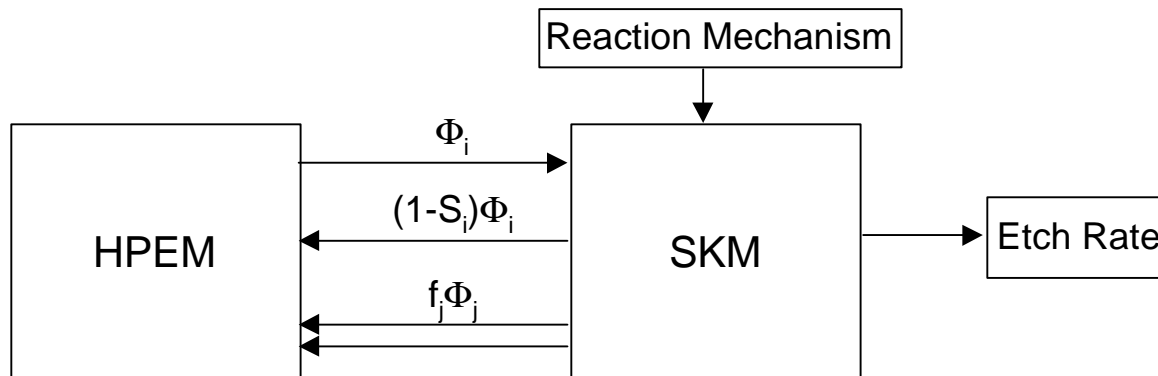
HYBRID PLASMA EQUIPMENT MODEL (HPEM)

- Modular simulator addressing low temperature, low pressure plasmas.
- EMM calculates electromagnetic fields and magneto-static fields.
- EETM computes electron impact source functions and transport coefficients.
- FKM derives the densities, momentum and temperature of plasma species.
- VPEM shell can be added to HPEM for process control.



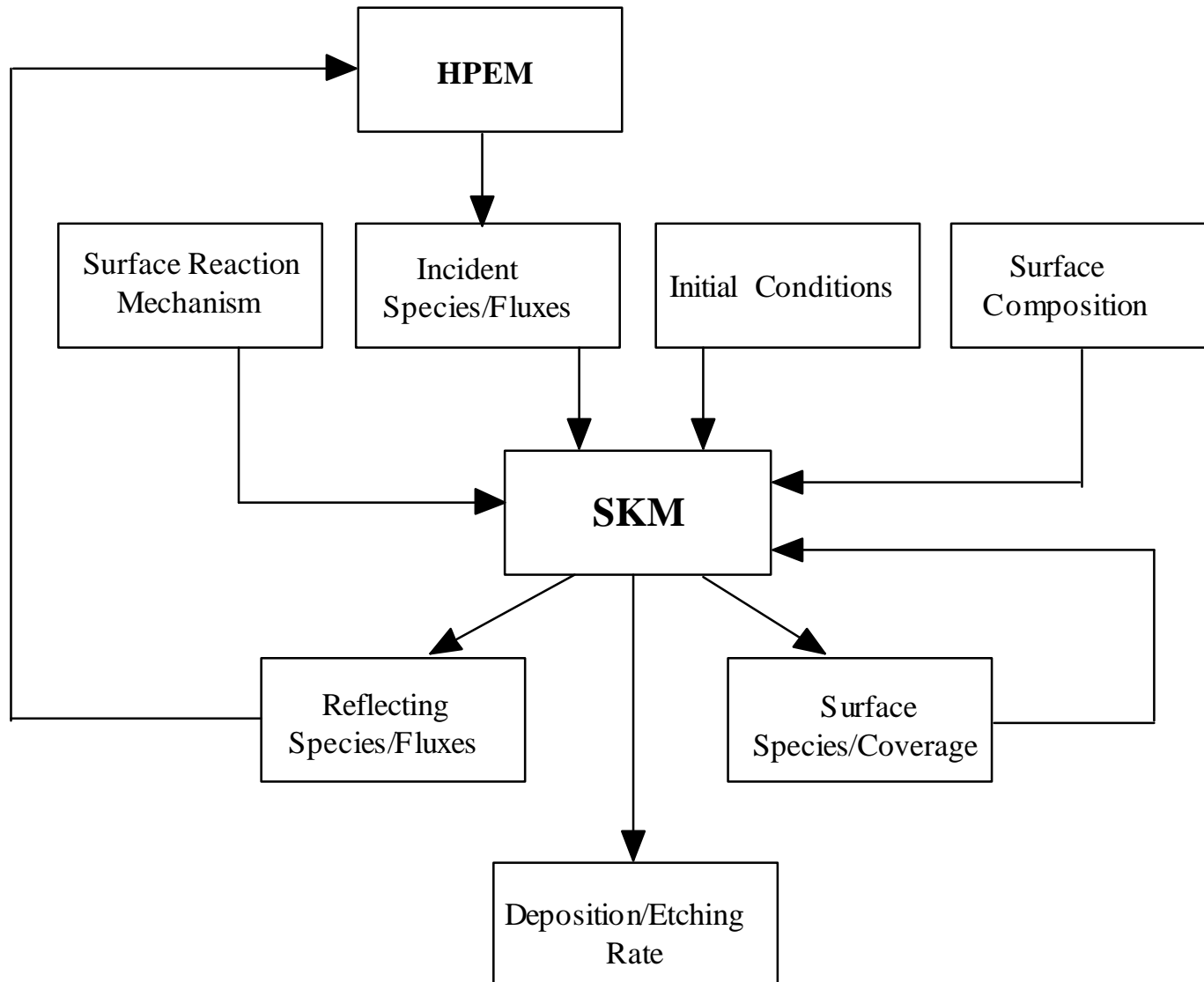
SURFACE KINETICS MODEL (SKM)

- The Surface Kinetics Model (SKM) is an integrated module of the HPEM.
- The SKM
 - uses reactant fluxes from the HPEM
 - applies a user defined reaction mechanism
 - updates surface sticking and product reflection coefficient for the HPEM
 - calculates surface coverages and etch rates



- The SKM is a multi-layer surface site balance model at every mesh point along the plasma-surface boundary.

C_mF_n PLASMA: SURFACE REACTIONS



ENERGY DEPENDENCE OF REACTION PROBABILITY

- All surface reactions involving ion reactants in the SKM allow for ion energy dependence.
- Ions are accelerated to the surface through the sheath, arriving on the surface with energy of

$$E_{\text{ion}} = Q f(r) V_{\text{sh}}(r)$$

where

Q: Ion charge.

f(r): Ratio of ion mean free path to sheath thickness.

$V_{\text{sh}}(r)$: Sheath voltage drop.

- Surface reactions have a general energy dependence given by

$$K = K_0 (E_{\text{ion}}^n - E_{\text{th}}^n) / (E_{\text{ref}}^n - E_{\text{th}}^n)$$

where

E_{ion} : Incident ion energy.

K_0 : Etching yield or reaction probability for ion with energy E_{ref} .

E_{th} : Threshold energy for the process.

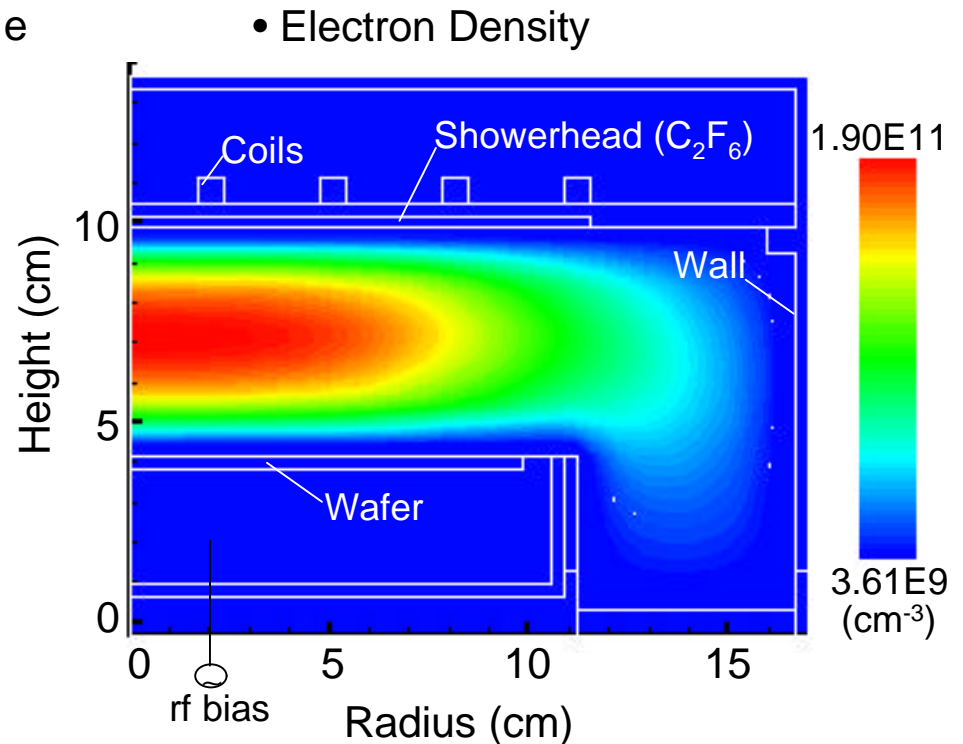
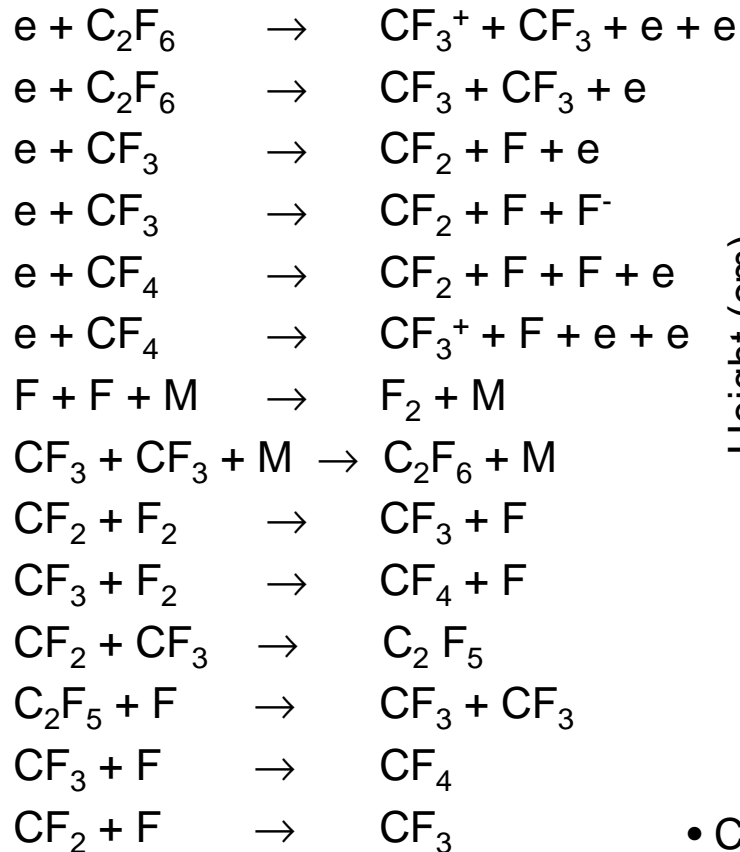
n: Energy dependence (1/2 for this work).

POLYMER LAYER PASSIVATION

- During Si/SiO₂ etching by fluorocarbon plasmas, a C_mF_n polymer layer will simultaneously deposit on the surface of the wafer and the reactor wall.
- The SKM allows the growth of the passivation layer by CF_x radical polymerization on the surface. F atoms etching and energetic ion sputtering can consume the layer.
- The steady state passivation thickness is reached when the film generation and consumption rates are balanced.
- SKM uses passivation thickness dependent rates for mass and energy transfers through the layer.

C₂F₆ ETCHING of Si

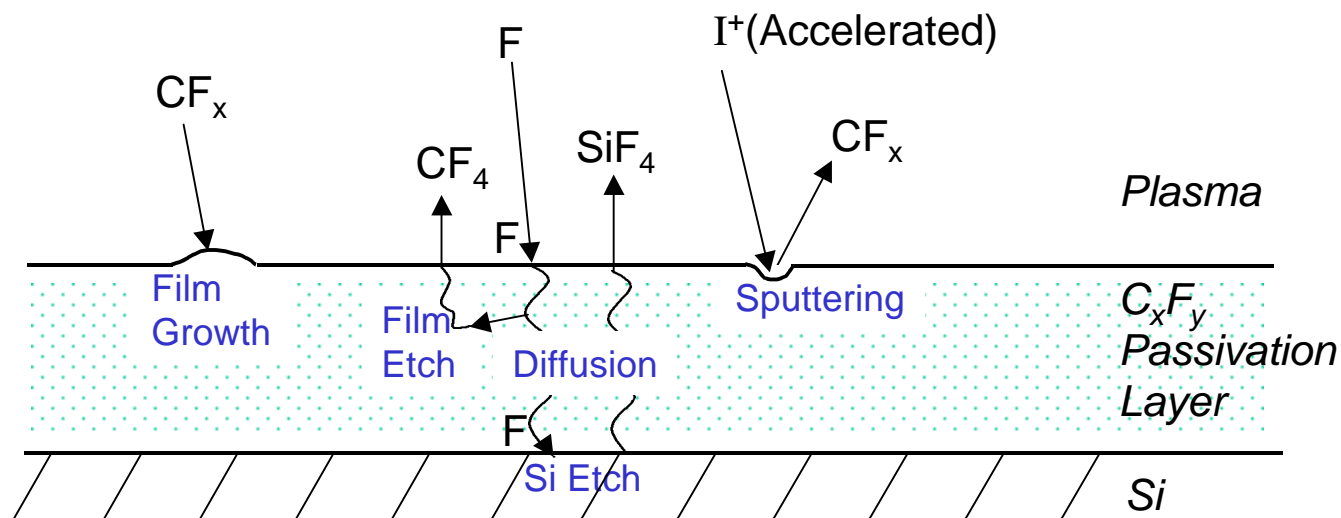
- Simulations of C₂F₆ etching of Si in an ICP reactor were performed.
- Representative gas phase reactions:



• C₂F₆, 10 mtorr, 200 sccm, 650 W ICP, 100 V bias

REACTION MECHANISM FOR C₂F₆ ETCHING OF Si

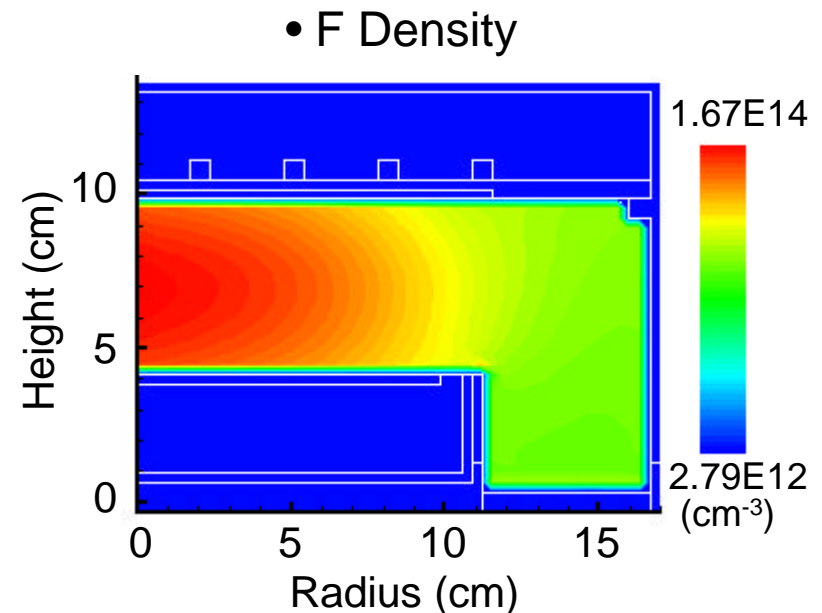
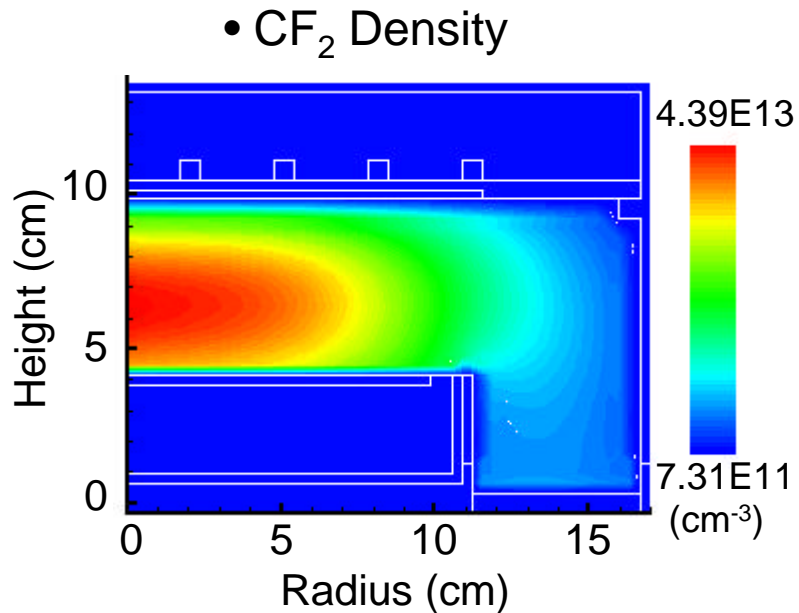
- The reaction mechanism was based on the work of G. S. Oehrlein *et al* *.
- A C_xF_y polymer layer is formed on the Si surface in coincidence with Si etching. The steady state passivation layer thickness is a balance of CF_x deposition, ion sputtering and F etching of the layer.
- Si etching precursor (F) needs to diffuse through the passivation layer.



* T.E.F.M. Standert, M.Scharkens, N.R.Rueger, P.G.M. Sebel, and G.S. Oehrlein, J. Vac. Sci. Technol. A 16(1), 239 (1998)

RADICAL DENSITIES: BASE CASE

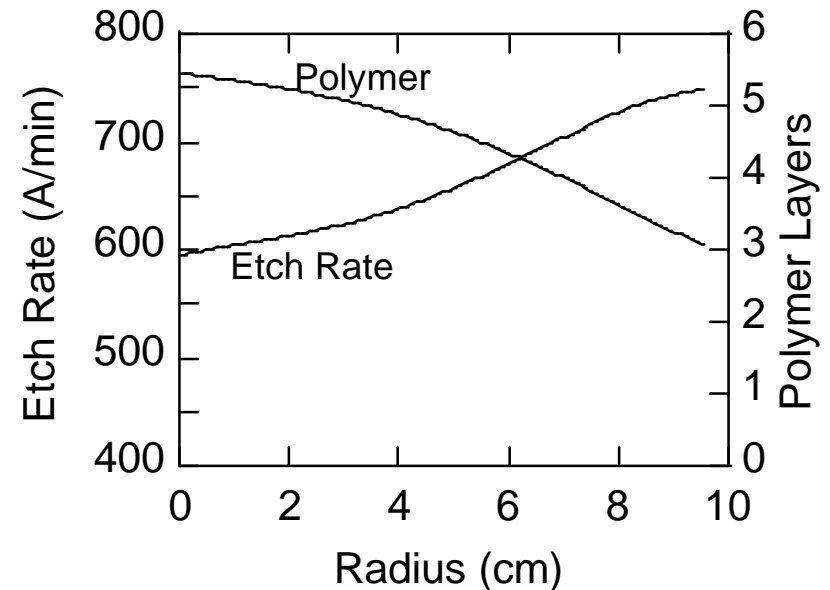
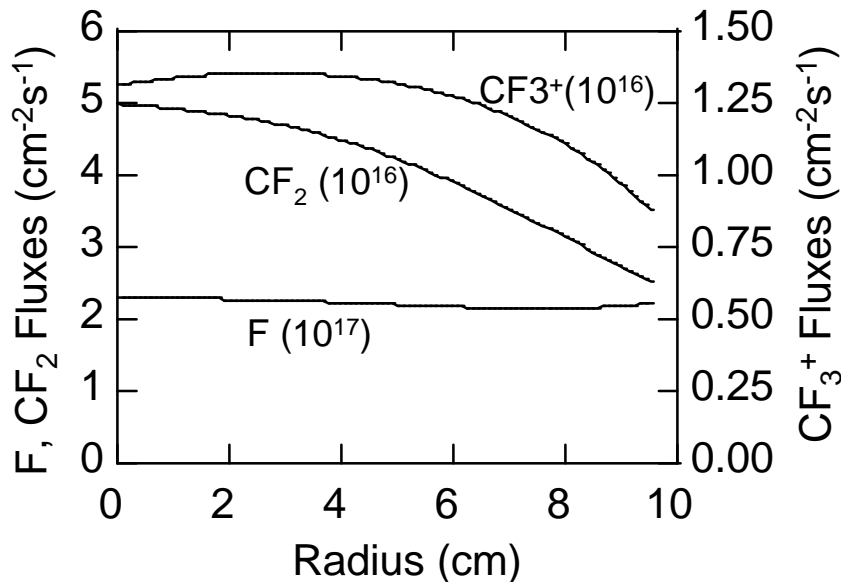
- Radical densities peak on the axis.
- The less uniform CF_2 density distribution is due to its higher loss rate at the reactor wall.



- C_2F_6 , 10 mtorr, 200 sccm, 650 W ICP, 100 V bias.

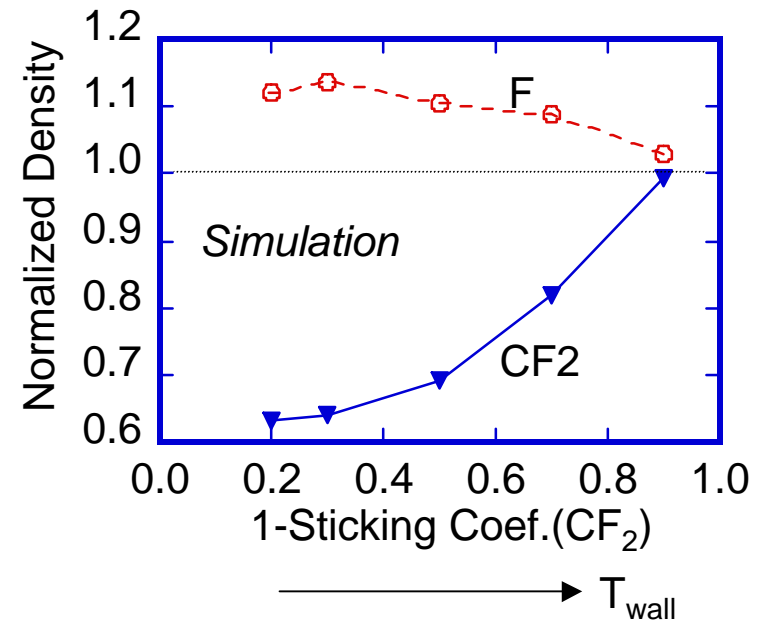
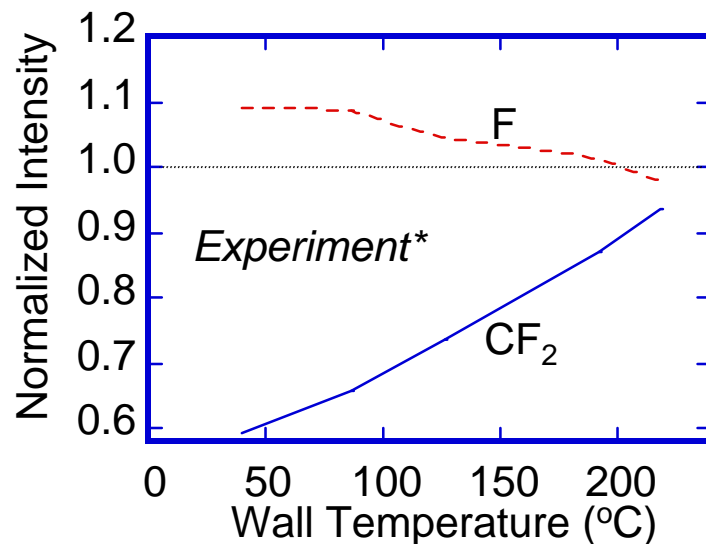
RADIAL DISTRIBUTIONS: FLUXES & ETCH RATE

- Radial flux distribution of CF_2 is less uniform than that for F as a consequence of the low sticking coefficient for F.
- Polymer thickness follows the CF_2 flux for this case.
- Etch rate increases in the radial direction, inversely to the polymer thickness due to F atom diffusion.



INFLUENCE OF WALL TEMPERATURE

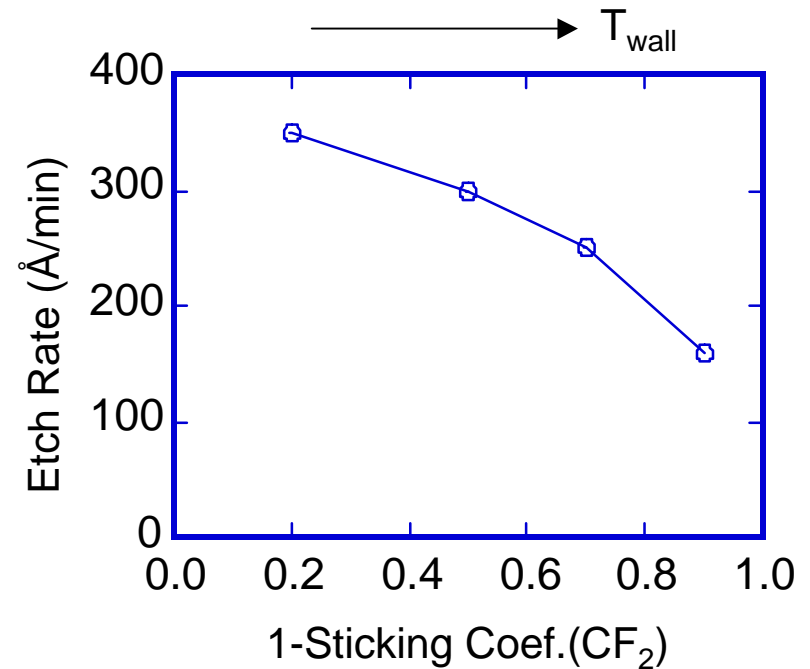
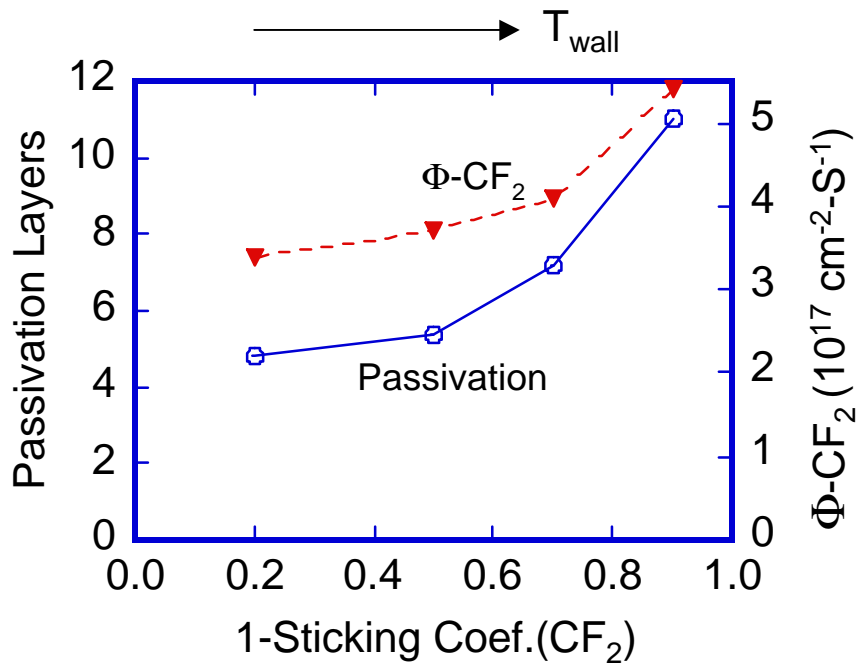
- Experiments showed a variation of radical densities vs. T_{wall} . * We simulated the consequence of varying T_{wall} by modifying the sticking coefficient of CF_2 to the wall.
- With increasing T_{wall} , the CF_2 loss rate is smaller due to the lower sticking coefficient, which produces an increase of CF_2 density in the bulk plasma.
- The resulting gas chemistry favors consumption of F atoms. So F density drops.



* M.Scharkens, R.C.M. Bosch, and G.S. Oehrlein,
J. Vac. Sci. Technol. A 16(1), 239 (1998)

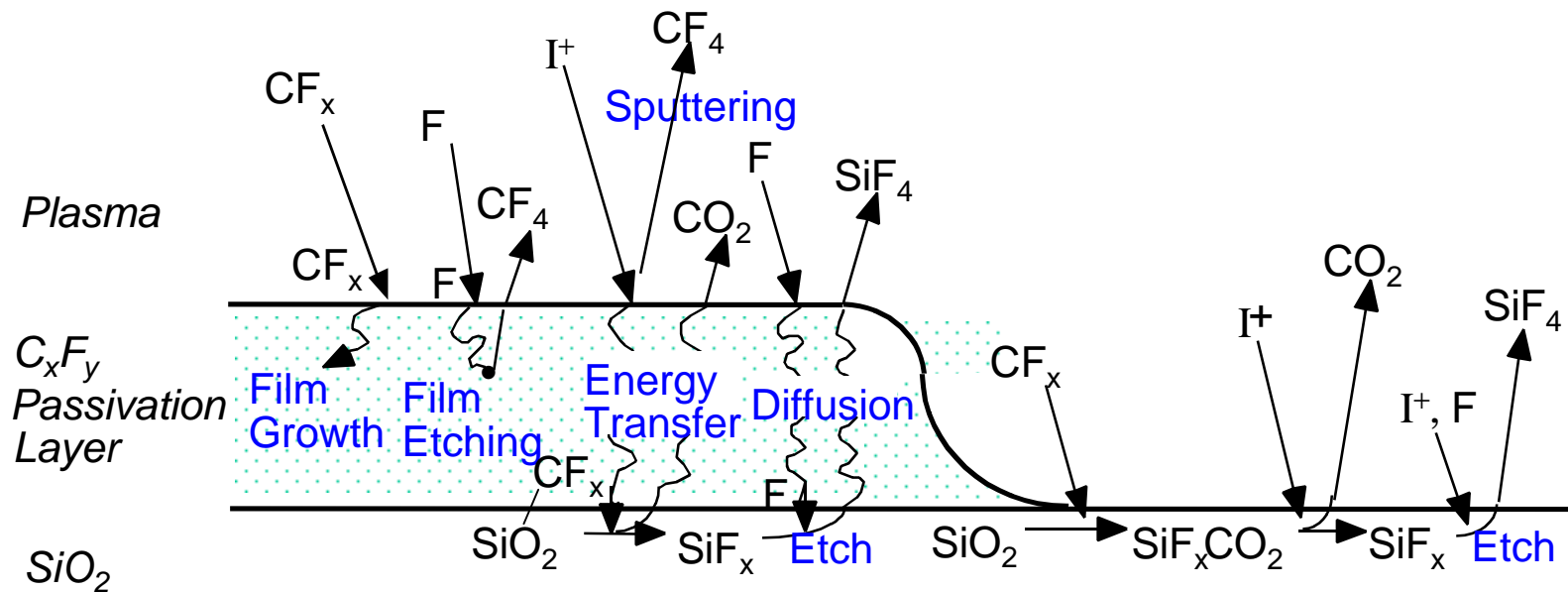
PASSIVATION LAYER AND ETCH RATE

- As wall temperature increases, the increased CF_2 density produces larger fluxes to the wafer. This leads to a thicker passivation layer.
- The diffusion flux of the Si etching precursor (F) through the passivation layer decreases with increasing passivation, so the etch rate drops.



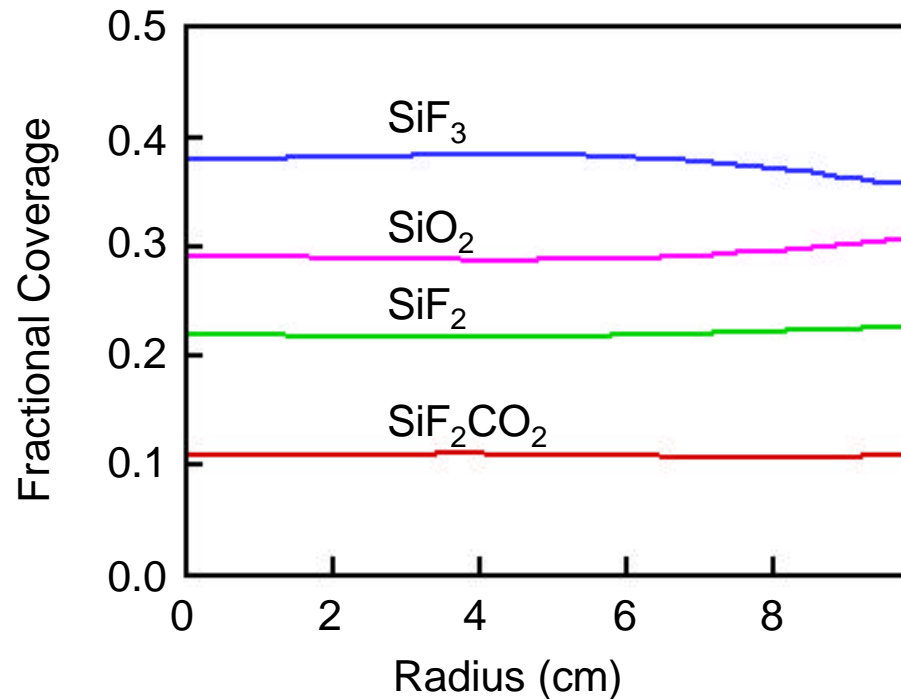
REACTION MECHANISM FOR C₂F₆ ETCHING OF SiO₂

- The reaction mechanism for SiO₂ etching is based on:
 - Growth of C_xF_y Passivation layer (balance of deposition and consumption).
 - Formation of complex at the interface between oxide and passivation layer resulting from chemisorption of CF_x.
 - Ion activated (through passivation layer) etching of complex. Rate of activation scales inversely with passivation layer thickness.
 - Diffusion of etch precursor and etch product.



SURFACE COVERAGE

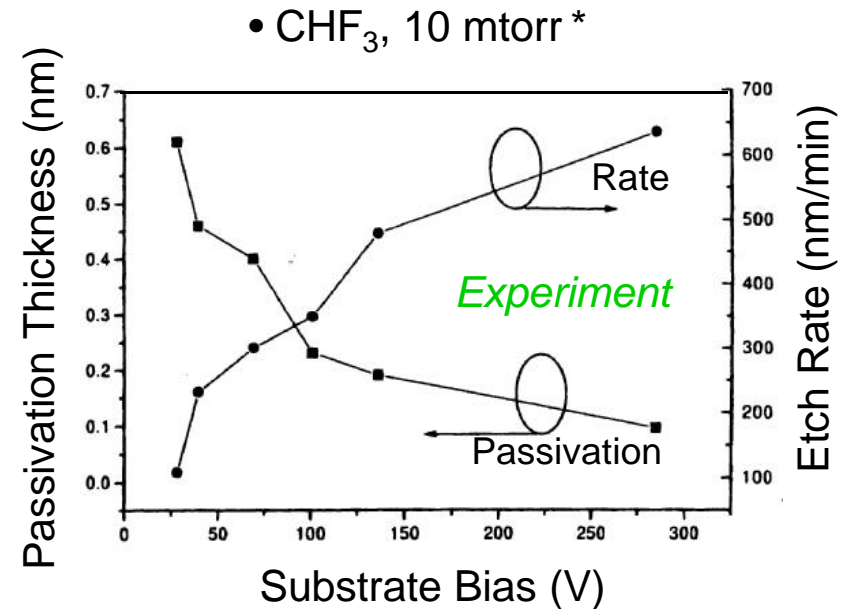
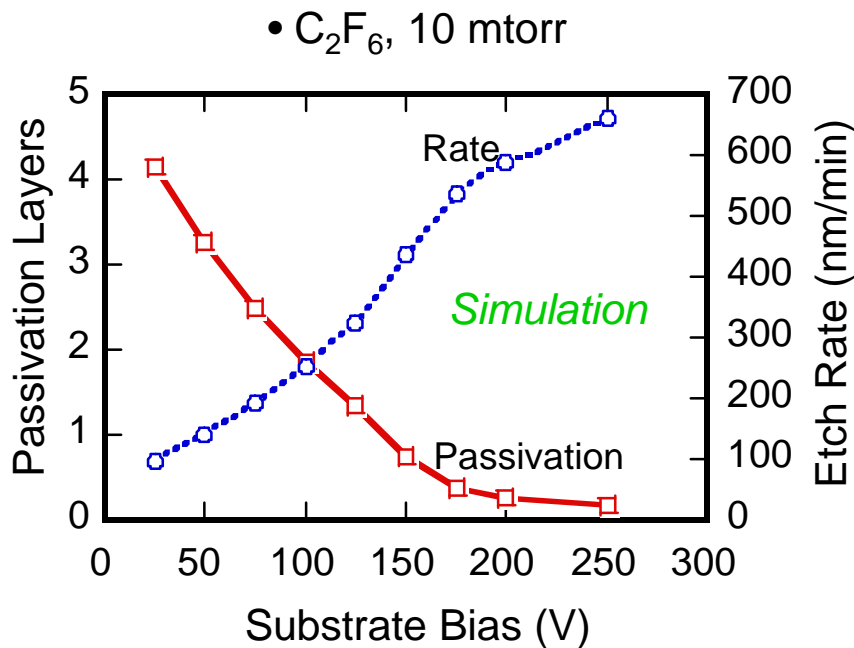
- The wafer surface sites are occupied by several surface species.
- The surface coverages at steady state depend on the relative rates of
 - complex formation
 - Ion activation
 - F atom etching
 - Sputtering



- C₂F₆, 10 mtorr
- 100 sccm
- 650 W ICP
- 100 V bias

SUBSTRATE BIAS

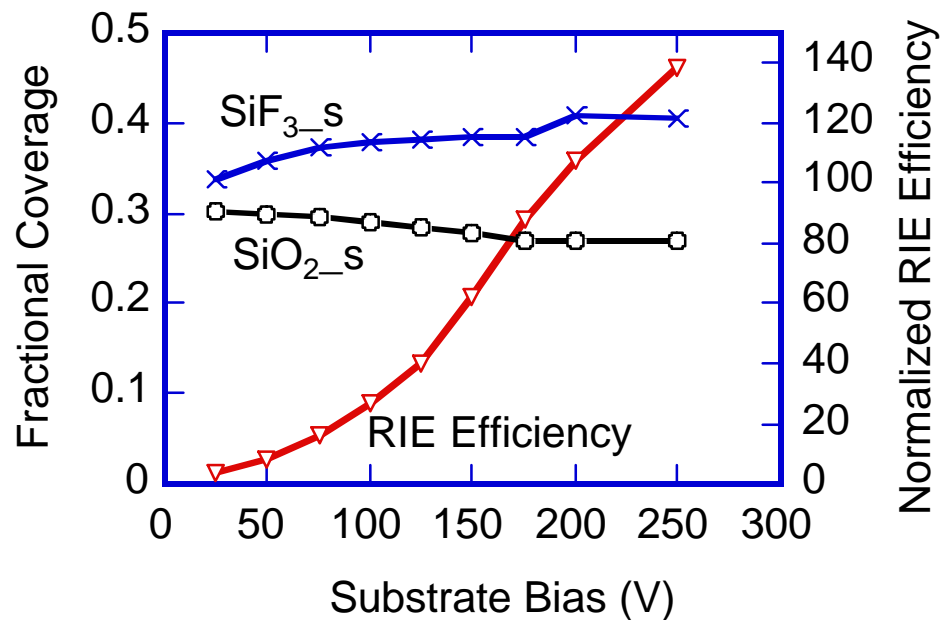
- With increasing substrate bias, the passivation layer thickness decreases and the etch rate increases.
- Simulations and experiments obtained similar trends.



*N.R. Rueger, G.S. Oehrlein et al, J. Vac. Sci. Technol. A 15, 1881 (1997)

SUBSTRATE BIAS

- With increasing bias, the variation in surface coverage is not big when the processes are not limited by the F atom diffusion.
- The normalized reactive ion etch efficiency increases with increasing substrate bias due to the increase in sheath voltage and decrease in passivation thickness.



Normalized RIE Efficiency

$$\propto \frac{V_{sheath}}{(1 + 0.3 \times [P - s])^2}$$

SUMMARY

- The Surface Kinetics Model (SKM) has been coupled with the HPEM to simulate surface interactions and their influences on the bulk plasma.
- In Si etching, higher loss of CF_2 on reactor walls leads to lower CF_2 density in the gas phase. This produces thinner passivation layers and higher etch rates.
- Consumption of C_xF_y at the polymer- SiO_2 interface during the SiO_2 etching leads to thinner passivation compared with Si etching.
- In SiO_2 etching, with increasing substrate bias, the surface coverages change little when the process is not limited by F diffusion.
- Increasing bias leads to higher sheath voltage and thinner passivation, so higher RIE efficiency. The etch rate increases accordingly.